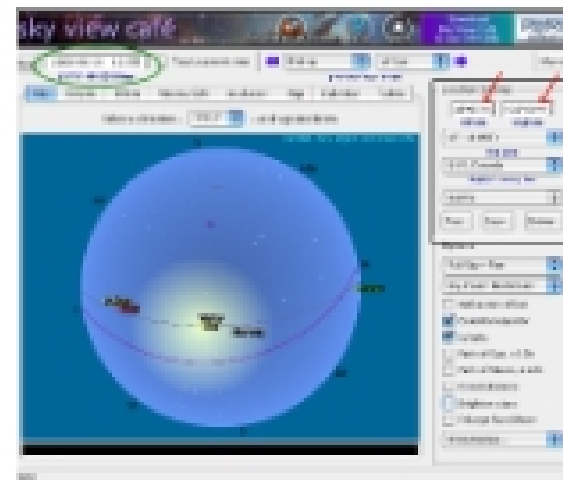


Announcements 10/03/11

- ✓ Check course calendar
 - Find link to lesson(s) for the week
 - Check out assignments
- ✓ Course Pack should be available before Thursday at UBS.
 - Spectral analysis (preparatory work + exercise)
 - Print your own copies for color
- ✓ Will get "real" star finder on Tuesday (please return card-stock ones to TA)
- ✓ Open House Jacobsen Observatory - Weds. 9 pm (7 pm is full) - if clear!

1

Loose ends - Sky View Cafe



2

Loose Ends - Quarks

MATTER AND ENERGY

Learning goals

- Distinguish among the following kinds of energy
 - kinetic
 - thermal
 - gravitational potential
 - electric potential
 - radiative
 - mass energy
- Give examples of how energy is conserved.

4

The Atom + isotopes & ions

atomic number = number of protons
atomic mass = number of protons + neutrons



atomic number = 1 atomic number = 2 atomic number = 6
atomic mass = 1 atomic mass = 4 atomic mass = 12
(1 electron) (2 electrons) (6 electrons)

The number of electrons in a neutral atom equals its atomic number.

What is an ion?

An atom that has lost 1 or more electrons

Isotopes of Hydrogen



Isotopes of Carbon



5

Nomenclature you need to know

Astronomers note how many electrons the atom has lost through Roman numerals.

H I = neutral hydrogen

He II = singly ionized helium

Ca II = singly ionized calcium

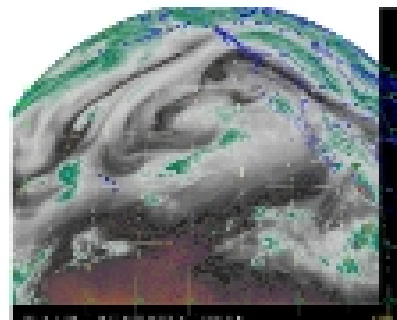
Fe IV = Iron atom has lost 3 electrons

Fe XII = Iron atom has lost ___?___ electrons

___?___ = oxygen atom missing 2 electrons

6

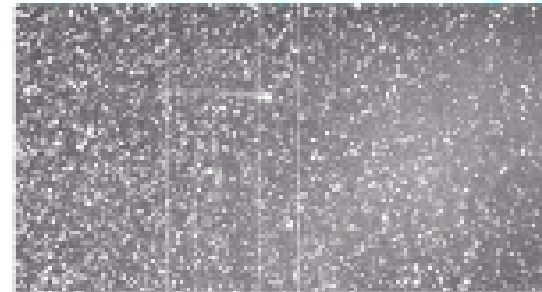
States of matter



7

Video taken by satellite STEREO

http://en.wikipedia.org/wiki/File:Encke_tail_rip_off.ogg



PLASMA

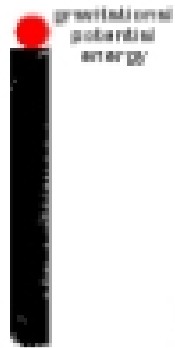
8

Energy

Kinetic vs Gravitational Potential

$KE = \frac{1}{2} mv^2$

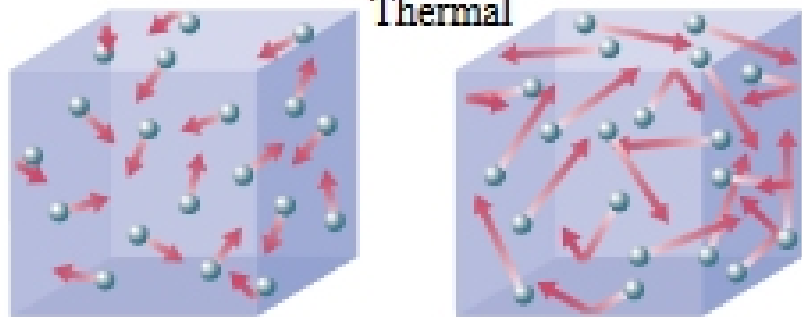
$GPE = mgh$



9

Energy

Thermal



Longer arrows mean higher average speed.


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Thermal energy is a measure of the average kinetic energy of a system of particles.


10

Energy

Radiative



Electric Potential



+

-

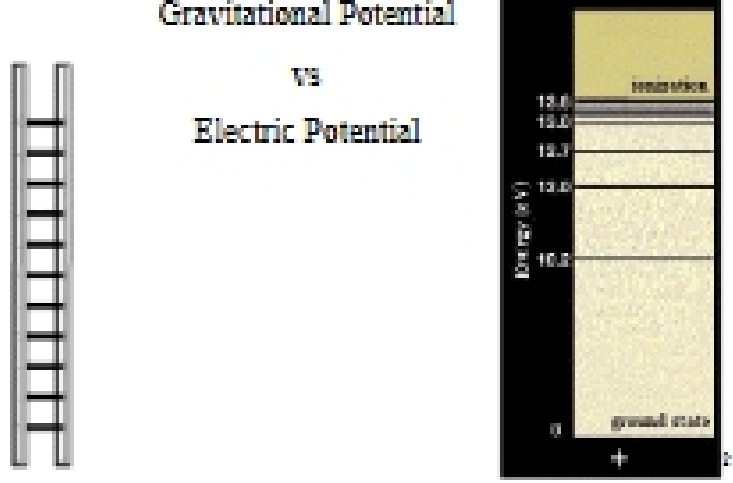
11

Energy

Gravitational Potential

VS

Electric Potential



12