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## Petrology & Petrography

**Petrology** - The branch of geology dealing with the origin, occurrence, structure, and history of rocks.

**Petrography** - The branch of geology dealing with the description and systematic classification of rocks, especially by microscopic examination of thin sections. Petrography is a subfield of Petrology.

In this course, most of the lecture material falls under the field of Petrology, while much of the laboratory material falls in the field of Petrography.

### Introduction to Igneous Rocks

An *igneous rock* is any crystalline or glassy rock that forms from cooling of a magma.

A *magma* consists mostly of liquid rock matter, but may contain crystals of various minerals, and may contain a gas phase that may be dissolved in the liquid or may be present as a separate gas phase.

Magma can cool to form an igneous rock either on the surface of the Earth - in which case it produces a *volcanic* or *extrusive igneous rock*, or beneath the surface of the Earth, - in which case it produces a *plutonic* or *intrusive igneous rock*.

### Characteristics of Magma

#### Types of Magma

Types of magma are determined by chemical composition of the magma. Three general types are recognized, but we will look at other types later in the course:

1. *Basaltic magma* -- SiO<sub>2</sub> 45-55 wt%, high in Fe, Mg, Ca, low in K, Na
2. *Andesitic magma* -- SiO<sub>2</sub> 55-65 wt%, intermediate in Fe, Mg, Ca, Na, K
3. *Rhyolitic magma* -- SiO<sub>2</sub> 65-75%, low in Fe, Mg, Ca, high in K, Na

#### Gases in Magmas

At depth in the Earth nearly all magmas contain gas dissolved in the liquid, but the gas forms a separate vapor phase when pressure is decreased as magma rises toward the surface. This is similar to carbonated beverages which are bottled at high pressure. The high pressure keeps the gas in solution in the liquid, but when pressure is decreased, like when you open the can or bottle, the gas comes out of solution and forms a separate gas phase that you see as bubbles. Gas gives magmas their explosive character, because volume of gas expands as pressure is

reduced. The composition of the gases in magma are:

- Mostly H<sub>2</sub>O (water vapor) with some CO<sub>2</sub> (carbon dioxide)
- Minor amounts of Sulfur, Chlorine, and Fluorine gases

The amount of gas in a magma is also related to the chemical composition of the magma. Rhyolitic magmas usually have higher dissolved gas contents than basaltic magmas.

### Temperature of Magmas

Temperature of magmas is difficult to measure (due to the danger involved), but laboratory measurement and limited field observation indicate that the eruption temperature of various magmas is as follows:

- Basaltic magma - 1000 to 1200°C
- Andesitic magma - 800 to 1000°C
- Rhyolitic magma - 650 to 800°C.

### Viscosity of Magmas

*Viscosity* is the resistance to flow (opposite of fluidity). Viscosity depends on primarily on the composition of the magma, and temperature.

- Higher SiO<sub>2</sub> (silica) content magmas have higher viscosity than lower SiO<sub>2</sub> content magmas (viscosity increases with increasing SiO<sub>2</sub> concentration in the magma).
- Lower temperature magmas have higher viscosity than higher temperature magmas (viscosity decreases with increasing temperature of the magma).

Thus, basaltic magmas tend to be fairly fluid (low viscosity), but their viscosity is still 10,000 to 100,000 times more viscous than water. Rhyolitic magmas tend to have even higher viscosity, ranging between 1 million and 100 million times more viscous than water. (Note that solids, even though they appear solid have a viscosity, but it is very high, measured as trillions time the viscosity of water). Viscosity is an important property in determining the eruptive behavior of magmas.

Summary Table					
Magma Type	Solidified Rock	Chemical Composition	Temperature	Viscosity	Gas Content
Basaltic	Basalt	45-55 SiO <sub>2</sub> %, high in Fe, Mg, Ca, low in K, Na	1000 - 1200 °C	10 - 10 <sup>3</sup> PaS	Low
Andesitic	Andesite	55-65 SiO <sub>2</sub> %, intermediate in Fe, Mg, Ca, Na, K	800 - 1000 °C	10 <sup>3</sup> - 10 <sup>5</sup> PaS	Intermediate
Rhyolitic	Rhyolite	65-75 SiO <sub>2</sub> %, low in Fe, Mg, Ca, high in K, Na.	650 - 800 °C	10 <sup>5</sup> - 10 <sup>9</sup> PaS	High

## Plutonic (Intrusive) Igneous Rocks

### Hypabyssal Intrusions

Intrusions that intrude rocks at shallow levels of the crust are termed hypabyssal intrusions. Shallow generally refers to depths less than about 1 km. Hypabyssal intrusions always show sharp contact relations with the rocks that they intrude. Several types are found:

- **Dikes** are small (<20 m wide) shallow intrusions that show a discordant relationship to the rocks in which they intrude. Discordant means that they cut across preexisting structures. They may occur as isolated bodies or may occur as swarms of dikes emanating from a large intrusive body at depth.
- **Sills** are also small (<50 m thick) shallow intrusions that show a concordant relationship with the rocks that they intrude. Sills usually are fed by dikes, but these may not be exposed in the field.
- **Laccoliths** are somewhat large intrusions that result in uplift and folding of the preexisting rocks above the intrusion. They are also concordant types of intrusions.

