

**Chemistry 506: Allied Health Chemistry 2****Chapter 13: Aldehydes and Ketones****Functional Groups with Double Bonds to Oxygen**

Introduction to General, Organic & Biochemistry, 5<sup>th</sup> Edition by  
Bettelheim and March: Chapter 13, Pages 425-450

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**Outline**

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**13A Section(s) 13.1/2/3 Introduction, Nomenclature, and Properties****□ Carbonyl Groups****□ C - O  $\pi$  bond and  $\sigma$ -bond****□ Relative Electronegativities of C and O****□ Polarity of Bond****□  $\delta^+$  charge on C****□  $\delta^-$  charge on O****□ Dipole - Dipole forces cause  $M_p$  and  $B_p$  increases****□ Lone Pairs on Oxygen**

- Aldehydes

- At least one H on Carbonyl Carbon

- Formaldehyde (Methanal)

- IUPAC Nomenclature

- anal ending

- Examples