

## Today's Learning Goals

- Mini-review: basic properties of stars
- Stellar Structure (The Sun is typical)
- How most stars derive their energies
- **The H-R Diagram** ← **IMPORTANT!**
  - The Main Sequence (93% of all stars)
  - Red Giants and white dwarfs (7%)
  - Correlations of properties
  - Observational clues for stellar evolution

## Review: Define \_\_\_\_

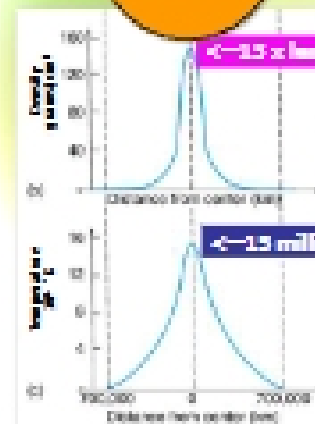
- (measurement units??)
- Brightness
  - Luminosity
  - Color
  - Temperature
  - Spectrum
  - Size
  - Mass and density
  - Gravity
  - 'Hydrostatic Equilibrium'

## Stellar Structure

- The internal structure of the Sun
- > What does 'structure' actually mean?
  - > The Earth's atmosphere is a good starting point



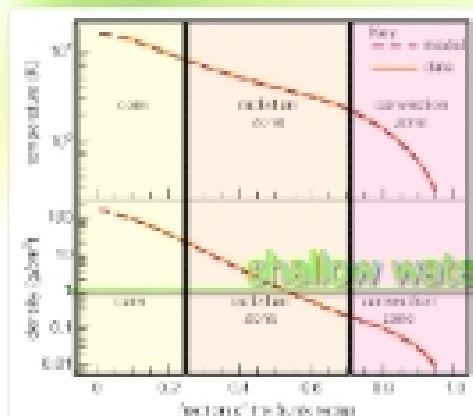
**Density & Temperature increase rapidly towards the core of the "bell"**



Why? The weight of the layers above compresses and heats the gas.

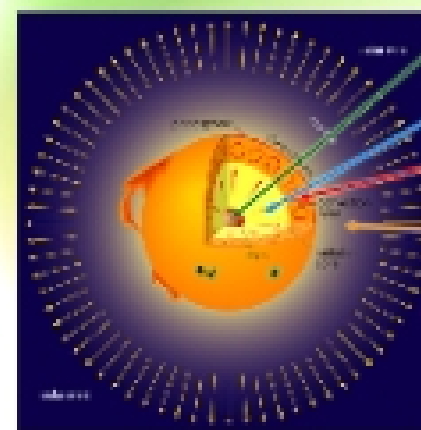
The weight of the Earth's atmosphere above you pushes on you @ 15.4 pounds per ft² at sea level (1/3 less @ 10,000 ft)

## How the Bell Tolls: results of physical models



**Fitting the model:** Vary the structure in the ringing regions of the model sun until agreement with the patterns of surface oscillations is reached. Then use the model to predict the density and temperature of the solar core.

## The Sun's major regions

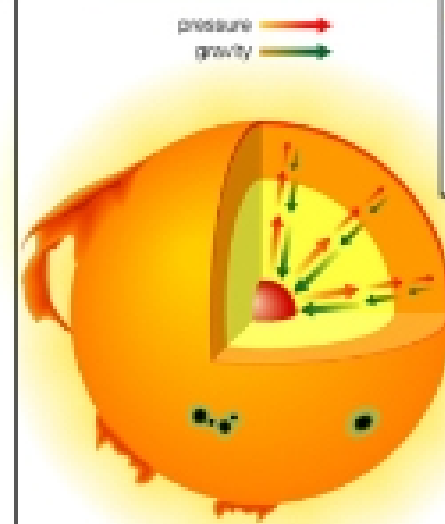


- **Core** (remember this!) (actual 150 g/cm³, **temp** 15 x 10⁶ K)
- **Radiative zone** (no circulation)
- **Convective zone** (vigorous circulation)
- **Corona** (hot plasma @ 10⁶ K)
- **Solar Wind:** A weak but steady flow of charged particles from the surface of the Sun punctuated by flares and CME's

## How does the Sun shine?



## The Fiery Core is a Star's Ultimate Heat Source



- Density of  $160 \text{ gm/cm}^3$ ;  $\sim 15$  times that of a lead brick;  $160 \times$  water.
- Fully ionized plasma - a dense gas of nuclei and their stripped off electrons

## What powers the Sun? - Chemical Energy? Is it on FIRE?

$\frac{\text{Chemical Energy Content}}{\text{Luminosity}} \sim 10,000 \text{ years}$

NO!

The Sun's chemical energy reserves would be consumed in  $10^4 \text{ y}$ .

## What powers the Sun? - Gravitational Energy? Is it CONTRACTING?

$\frac{\text{Gravitational Potential Energy}}{\text{Luminosity}} \sim 25 \text{ million years}$

NO!

Although gravity compresses and heats the solar interior, light energy emitted at the surface soon sheds this heat, and the Sun would collapse to a black hole in  $\approx 10^8 \text{ y}$ .

## What powers the Sun? — Nuclear Energy?

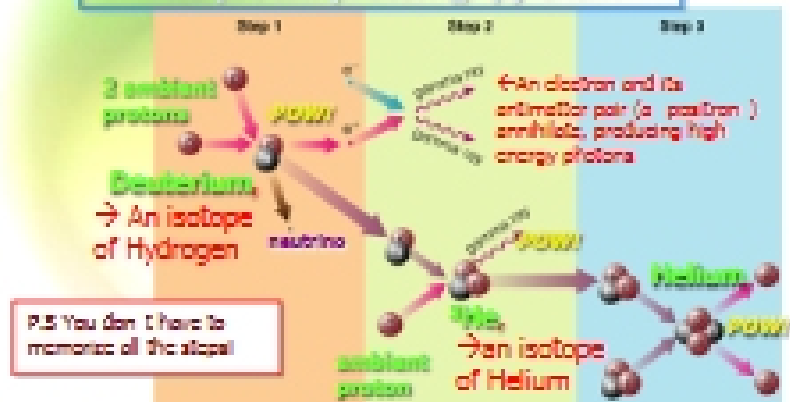


4 Hydrogen Atoms fuse to make 1 Helium Atom and a bunch of energy.

As we're about to see, the energy reserves will last  $10^{10} \text{ y}$ . So this looks like a sensible answer!

## Energy Production in Stars

### The proton-proton (p-p) chain



Net: In = 6 protons, Out = 1 He<sup>4</sup> + 2 protons + e<sup>+</sup> + light

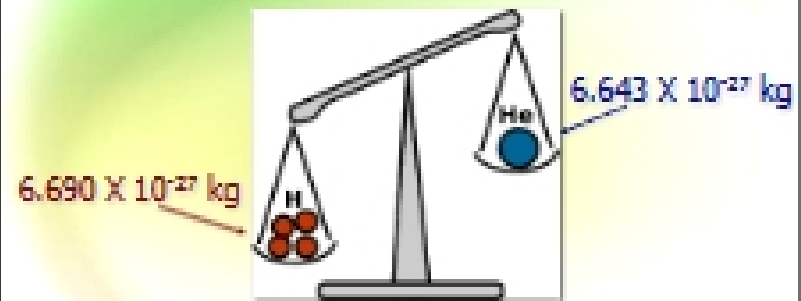
## Why extra energy at the end?



$$E = Mc^2!!!$$

Some mass converts to energy!

## How much mass per reaction?



Mass Difference =  $0.047 \times 10^{-27}$  kg  
(0.7% of the original mass)

Energy released per reaction = mass lost  $\times c^2$

## How much energy?

Doesn't take much mass!



Two 200-lb jocks converted to pure energy!

Releases  $2 \times 10^{28}$  ergs, which is  $\sim 4 \times 10^{12}$  kW hrs, or  $\frac{1}{5}$  of the entire US energy consumption per year.

## How much energy?

- The fusion process used by the Sun isn't 100% efficient - just 0.7%
  - (99.3% of mass remains after a helium nucleus is produced)
- 10% of the Sun's mass is in the core where combustion occurs
  - ...so 0.07% of the Sun's mass of hydrogen will ultimately be consumed!
  - Nonetheless, that's lots of energy

How long will the energy reserves last?  
Answer:  $\approx 10^{10}$  years ... let's do the math

## How much mass EVERY SECOND?

1 Solar Luminosity =  $3.8 \times 10^{26}$  watts  
1 watt =  $10^7$  erg per second

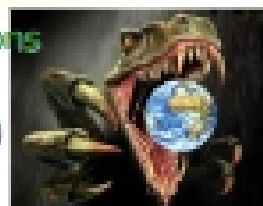
In one second, Sun releases  $\sim 4 \times 10^{33}$  erg, so:

Energy released = (mass destroyed)  $c^2$ , so:

Mass destroyed =  $(4 \times 10^{33} \text{ erg}) / (3 \times 10^{10} \text{ cm/s})^2$

=  $4 \times 10^{12}$  grams = 4 million tons is consumed every second!

(1.3 Earth masses since the dinosaurs)



## How long will the Fuel last?

- Lifetime =  $\frac{\text{energy reserves}}{\text{rate of energy consumption}}$
- Example: a car with a 10-gallon tank burning 2 gallons/hour can run for 5 hours
- Sun: energy reserves
  - = 0.7%  $\times$  core mass of hydrogen  $\times c^2$
  - =  $0.007 \times [0.1 \times 2 \times 10^{33} \text{ gm}] \times c^2$
  - =  $1.4 \times 10^{33} \text{ gm} \times c^2 = 1.3 \times 10^{51} \text{ ergs}$
- Sun: rate of consumption = luminosity
  - =  $4 \times 10^{33} \text{ erg/sec}$



The fuel will last  $1.3 \times 10^{51} \text{ ergs} / 4 \times 10^{33} \text{ erg/sec} = 3.1 \times 10^{17} \text{ sec} = 10^{10} \text{ years!}$