

## Rotation of Membrane Proteins : Determining Correlation Time

- Fluorescence/Phosphorescence
- Saturation Transfer EPR

## Hydrophobicity Scales of Amino Acids:

Wimley and White Hydrophobicity Measurements

3 papers:

Anal Biochem.213, 213-217 (1993)

Biochemistry 35, 5109-5124 (1996)

Nature Struc Bio 3, 842-848 (1996)

Fluorescence anisotropy measurements are useful for labeled proteins in solution, but what about proteins imbedded into the bilayer? Recall: Rotational diffusion can only be measured if a significant amount of rotation occurs during the lifetime of the excited state. Membrane imbedded proteins have correlation times as slow as  $\mu\text{s}$  to  $\text{ms}$ .

~ns

**Table 8-3**  
Typical fluorescent probes

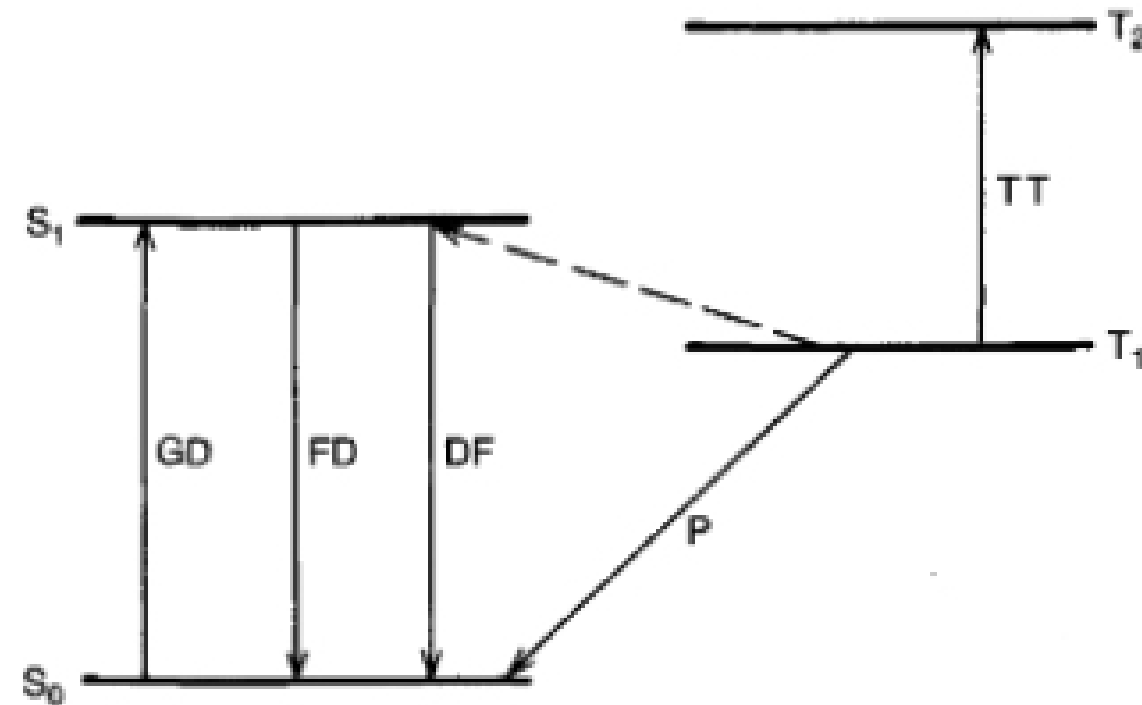
Probe*	Uses	Absorption		Emission <sup>†</sup>			Sensitivity
		$\lambda_{\text{max}}$ (nm)	$\epsilon_{\text{max}}$ $\times 10^{-3}$	$\lambda_{\text{max}}$ (nm)	$\phi_F$	$\tau_F$ (nsec)	$\epsilon_{\text{max}}\phi_F$ $\times 10^{-2}$
Dansyl chloride	Covalent attachment to protein: Lys, Cys	330	3.4	510	0.1	13	3.4
1,5-I-AEDANS	Covalent attachment to protein: Lys, Cys	360	6.8	480	0.5	15	34
Fluorescein isothiocyanate (FITC)	Covalent attachment to protein: Lys	495	42	516	0.3	4	116
8-Anilino-1-naphthalene sulfonate (ANS)	Noncovalent binding to proteins	374	6.8	454	0.98	16	67
Pyrene, and various derivatives	Polarization studies on large systems	342	40	383	0.25	100	100
Ethenoadenosine, and various derivatives	Analogues of nucleotides bind to proteins, incorporate into nucleic acids	300	2.6	410	0.40	26	10
Ethidium bromide	Noncovalent binding to nucleic acids	515	3.8	600	~1	26.5	38
Proflavine monosemicarbazide	Covalent attachment to RNA 3'-ends	445	15	516	0.02	—	30

<sup>†</sup> Values shown for  $\phi_F$  and  $\tau_F$  are near the maximum typically observed in biological samples at ambient temperature. Other (considerably smaller) values often are found.

\* Structures of these probes are shown in Figure 8-16.

Possible solutions to measuring slower correlation times:

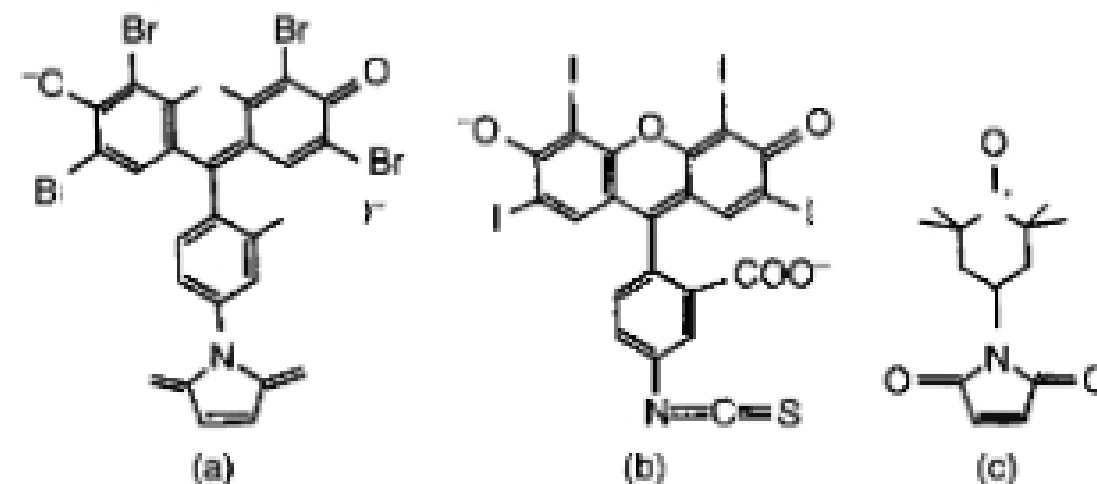
- 1) Utilize the triplet state (optical spectroscopy)
- 2) Saturation transfer EPR



**FIGURE 11.2** Electronic energy level diagram illustrating different methods of detecting molecules in the triplet state T<sub>1</sub>. Phosphorescence (P) is emission accompanying a radiative transition back to the ground state S<sub>0</sub> via the excited singlet state S<sub>1</sub>. Population of T<sub>1</sub> depletes the ground state, resulting in loss of intensity of singlet-singlet absorption (GD = ground state depletion) and prompt fluorescence (FD = fluorescence depletion) under continuous illumination. Finally, molecules in the triplet state undergo triplet-triplet absorption (TT) when illuminated with light of appropriate wavelength. T<sub>1</sub> is populated by intersystem crossing from S<sub>1</sub> after exciting the S<sub>0</sub>-S<sub>1</sub> transition. For simplicity, vibrational levels are omitted.

Rotational Diffusion constant for protein in membrane

$$D_R = kT / (4\pi a^2 h \eta)$$



**FIGURE 11.3** Commonly used probes for rotational diffusion measurements: (a) eosin-5-maleimide and (b) erythrosine-5-isothiocyanate are employed for optical methods, and the spin label (c) 6-MSL (N-[1-oxyl-2,2,6,6-tetramethyl-4-piperidinyl]maleimide) is employed for ST-EPR.