

**MSE 230
Fall 2012
EXAM I**

**Closed book
(1 sheet of notes, 8.5"x11" writing on both sides, allowed)**

Name: _____

Recitation (day and time): _____

Kara Luitjohan

R3 Thur. 10:30

R6 Thur. 4:30

Andrew Kustas

R10 Thur. 9:30

R5 Fri. 10:30

Abhishek Bawiskar

R4 Thur. 1:30

R12 Fri. 11:30

Yuefeng Wang

R8 Thur. 2:30

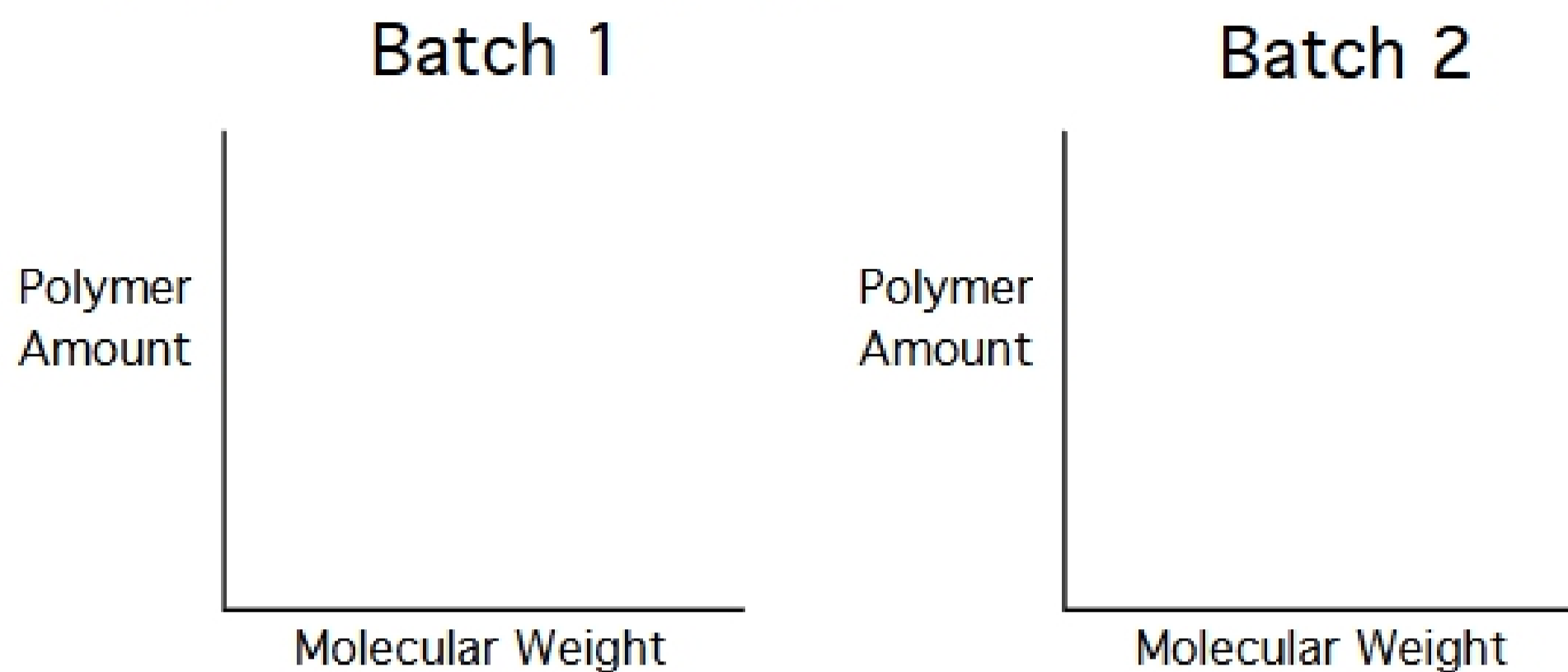
R11 Thur. 3:30

Good Luck!

1	a)	/05
	b)	/05
	c)	/10
	<i>total</i>	/20
2		
	<i>total</i>	/15
3		
		/20
4		
		/05
5	a)	/05
	b)	/10
	<i>total</i>	/15
Total		/75

(20) 1. Short answer questions.

(5) a) Make a qualitative sketch of the molecular weight distributions for two batches of polyethylene that were polymerized under different conditions. Batch 1 was polymerized such that $\bar{M}_w \gg \bar{M}_n$ and Batch 2 was polymerized such that $\bar{M}_w \approx \bar{M}_n$.



(5) b) Match the values of Young's modulus and coefficient of thermal expansion with the appropriate material.

Values of Young's modulus (GPa):

2.7, 115, 304

Coefficient of Thermal Expansion Values ($^{\circ}\text{C}^{-1}$):

3.0×10^{-6} , 17×10^{-6} , 120×10^{-6}

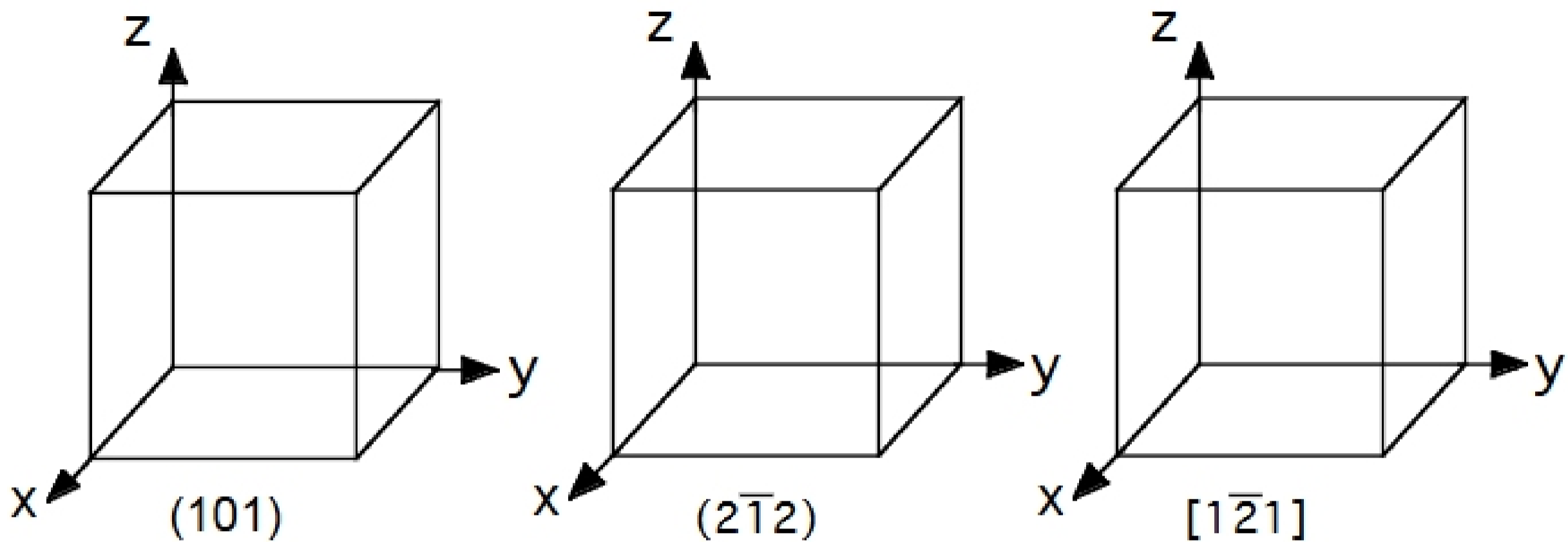
Material	E (GPa)	$\alpha(^{\circ}\text{C}^{-1})$
Copper		___ $\times 10^{-6}$
Polystyrene		___ $\times 10^{-6}$
Silicon Nitride		___ $\times 10^{-6}$

(10) c)

Circle either TRUE or FALSE

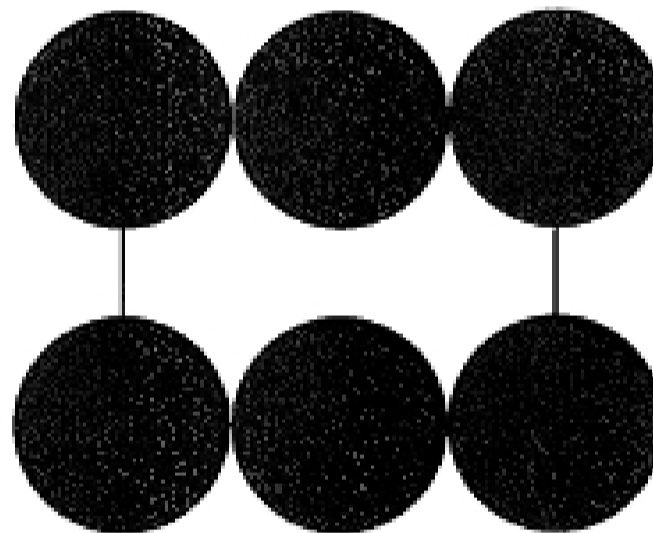
- | | | |
|---|------|-------|
| i: In general, volume is not conserved during elastic deformation. | TRUE | FALSE |
| ii: Given the same molecular weight and chemical composition, a polymer composed of linear chains will crystallize more readily than a polymer composed of branched chains. | TRUE | FALSE |
| iii: Heating a crystalline metal will cause its x-ray diffraction peaks to shift slightly to the right. | TRUE | FALSE |
| iv: The slope at $r=r_0$ of the potential energy vs. separation distance curve is proportional to Young's modulus. | TRUE | FALSE |
| v: In a BCC crystal structure the atoms touch along the $\langle 111 \rangle$. | TRUE | FALSE |

(15) 2. Crystals Structures, Planes and Directions

(6) a) Draw the following **inside** the cubes provided.

(4) b) Please fill in the blanks.

The drawing below shows the arrangement of atoms on the {_____} of a _____ metal.

(5) c) Which of the following are possible slip systems for a FCC metal? Circle your answer(s).

i) $(1\bar{1}1)[10\bar{1}]$

ii) $(1\bar{1}0)[110]$

iii) $(11\bar{1})[\bar{1}01]$

iv) $(1\bar{1}1)[0\bar{1}\bar{1}]$

v) $(10\bar{1})[\bar{1}1\bar{1}]$