

Physics for Scientists and Engineers I

PHY 2048, Section 4

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Chapter 0 - Introduction

- I. General
- II. International System of Units
- III. Conversion of units
- IV. Dimensional Analysis
- V. Problem Solving Strategies

I. Objectives of Physics

- Find the limited number of fundamental laws that govern natural phenomena.
- Use these laws to develop theories that can predict the results of future experiments.
- Express the laws in the language of mathematics.
- Physics is divided into six major areas:
 1. Classical Mechanics (PHY2048)
 2. Relativity
 3. Thermodynamics
 4. Electromagnetism (PHY2049)
 5. Optics (PHY2049)
 6. Quantum Mechanics

II. International System of Units

QUANTITY	UNIT NAME	UNIT SYMBOL
Length	meter	m
Time	second	s
Mass	kilogram	kg
Speed		m/s
Acceleration		m/s ²
Force	Newton	N
Pressure	Pascal	Pa = N/m ²
Energy	Joule	J = Nm
Power	Watt	W = J/s
Temperature	Kelvin	K

POWER	PREFIX	ABBREVIATION
10 ²³	peta	P
10 ²²	tera	T
10 ⁹	giga	G
10 ⁶	mega	M
10 ³	kilo	k
10 ²	hecto	h
10 ¹	deka	da
10 ⁰	deci	D
10 ⁻²	centi	c
10 ⁻³	milli	m
10 ⁻⁶	micro	μ
10 ⁻⁹	nano	n
10 ⁻¹²	pico	p
10 ⁻¹⁵	femto	f

III. Conversion of units

Chain-link conversion method: The original data are multiplied successively by conversion factors written as unity. Units can be treated like algebraic quantities that can cancel each other out.

Example: 316 feet/h \rightarrow m/s

$$\left(316 \frac{\cancel{\text{feet}}}{\cancel{\text{h}}}\right) \cdot \left(\frac{1 \cancel{\text{h}}}{3600 \text{s}}\right) \cdot \left(\frac{1 \text{ m}}{3.28 \cancel{\text{feet}}}\right) = 0.027 \text{ m/s}$$

IV. Dimensional Analysis

Dimension of a quantity: indicates the type of quantity it is; **length [L]**, **mass [M]**, **time [T]**

Dimensional consistency: both sides of the equation must have the same dimensions.

Example: $x = x_0 + v_0 t + at^2/2$

$$[L] = [L] + \frac{[L]}{[T]} [T] + \frac{[L]}{[T^2]} [T^2] = [L] + [L] + [L]$$

Note: There are no dimensions for the constant (1/2)

Table 1.6

Units of Area, Volume, Velocity, Speed, and Acceleration				
System	Area (L ²)	Volume (L ³)	Speed (L/T)	Acceleration (L/T ²)
SI	m ²	m ³	m/s	m/s ²
U.S. customary	ft ²	ft ³	ft/s	ft/s ²

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Significant figure \rightarrow one that is reliably known.

Zeros may or may not be significant:

- Those used to position the decimal point are not significant.
- To remove ambiguity, use scientific notation.

Ex: 2.56 m/s has 3 significant figures, 2 decimal places.
 0.000256 m/s has 3 significant figures and 6 decimal places.
 10.0 m has 3 significant figures.
 1500 m is ambiguous \rightarrow 1.5×10^3 (2 figures), 1.50×10^3 (3 fig.), 1.500×10^3 (4 figs.)

Order of magnitude \rightarrow the power of 10 that applies.