

GEOLOGY TEST 1 STUDY GUIDE

MINERALS

8 most abundant elements:	Common ionic form:	Relative Ionic Size:	Coordination #:
Oxygen (O)	O ⁻²	1.4	depends on size
Silicon (Si)	Si ⁺⁴	.26	4
Aluminum (Al)	Al ⁺³	.39	4 or 6
Iron (Fe)	Fe ⁺²	.63	6
Calcium (Ca)	Ca ⁺²	1.00	8
Sodium (Na)	Na ⁺¹	.99	8
Potassium (K)	K ⁺¹	1.37	8 or 12
Magnesium (Mg)	Mg ⁺²	.72	6

Crystalline vs Non-crystalline:

A crystalline solid has its molecules (or atoms) arranged in a lattice. That is, they are arranged in a regular fashion with equal spacing and angular relationships in all three directions.

A non-crystalline solid, also called amorphous, has no such regular arrangement of molecules.

Silicate Minerals:

olivines - $(Mg,Fe)_2SiO_4$; independent tetrahedra

pyroxenes - $(Mg,Fe)_2Si_2O_6$; single chain

amphiboles - $(W,X,Al)_{7-8}(Z_4O_{11})_2(OH)_2$; double chains

biotite mica - $K(Mg,Fe)_3(AlSi_3O_{10})(OH)_2$; sheet silicates

muscovite mica - $KAl_2(AlSi_3O_{10})(OH)_2$; sheet silicates

alkali feldspars - $KAlSi_3O_8$ --- $NaAlSi_3O_8$ (solid solution series) ; framework silicates

plagioclase feldspars - $NaAlSi_3O_8$ --- $CaAl_2Si_2O_8$ (solid solution series) ; framework silicates

quartz - SiO_2 ; framework silicates

Mafic- rocks low in silicon

Felsic- rocks high in silicon

EARTH'S HEAT

Heat Transfer:

Radiation- Electromagnetic waves that directly transport ENERGY through space.

Conduction- transfer of energy through matter from particle to particle

Convection- transfer of heat by the actual movement of the warmed matter.

Origin of Earth's Heat:

comets, asteroids, and other smaller masses

Geothermal Gradient: rate at which the temperature increases with depth in the Earth

Geobarometric Gradient:

MAGMAS

Melting of Silicate rocks: Volatiles (water)- Temperatures of minerals are reduced under high water pressure. Wet rocks melt at lower temps than do dry rocks.

Factors influencing viscosity- silica content and temperature.

Crystallization of magmas:

Fractional Crystallization- Rapid cooling, crystal settling or mineral overgrowth inhibiting diffusion prevents melt reacting with crystals

Equilibrium Crystallization- slow cooling allows previously formed crystals to react with melt to bring them into equilibrium as melt composition changes

Magma rise to shallow crustal levels:

1. The most obvious is an increase in temperature.
2. The introduction of water into rock. Water lowers the melting point of rocks.
3. A decrease in pressure of rocks that are already hot will cause them to melt.

Magma originate: crust or mantle

IGNEOUS ROCKS

Volcanic rocks (small crystals) – basalt- mafic, fine grained, dark color, andesite- intermediate, fine grained, rhyolite- felsic, fine grained, light color

Plutonic rocks (large crystals) – peridotite- ultramafic, coarse grained, gabbro- mafic, coarse grained, dark colored, diorite- intermediate, coarse grained, granite- felsic, coarse grained, light colored

Textures of rocks-

Batholith- a very large igneous intrusion extending deep in the earth's crust.

Dike- a long wall built to prevent flooding from the sea

Extrusive- igneous rocks that have been forced out in a molten or plastic condition upon the surface of the earth.

Intrusive- having been forced while in a plastic state into cavities or between layers

Pluton- body of intrusive igneous rock that is crystallized from magma slowly cooling below the surface of the Earth.

VOLCANOES

Types of Volcanoes:

Volcanic fields- small eruptions occur over a wide area

Cone Volcano- succession of small moderate eruptions from one location

Caldera Volcano- infrequent but moderate large eruptions

Cinder Volcano- steep conical hill with straight sides, explosive

Shield Volcano- very gentle slopes, convex upward, quiet

Strato Volcano- gentle lower slopes, steep upper slopes, concave upward, explosive

Volcanic Eruptions: